

PCTWORLD INTELLECTUAL PROPERTY ORGANIZATION
International Bureau

INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(51) International Patent Classification ⁶: B29C 55/02, B32B 27/30, 27/08	A1	(11) International Publication Number: WO 99/62694 (43) International Publication Date: 9 December 1999 (09.12.99)
(21) International Application Number: PCT/EP99/03758 (22) International Filing Date: 28 May 1999 (28.05.99) (30) Priority Data: 98401321.9 3 June 1998 (03.06.98) EP (71) Applicant (for all designated States except US): E.I. DU PONT DE NEMOURS AND COMPANY [US/US]; 1007 Market Street, Wilmington, DE 19898 (US). (72) Inventors; and (75) Inventors/Applicants (for US only): ANDREIS, Christine [DE/DE]; Warsbergerstrasse 1b, D-54457 Wincheringen (DE). FEYDER, Gusti [LU/LU]; 159, rue des Pommiers, L-2343 Luxembourg (LU). (74) Agents: POCHART, François et al.; Cabinet Hirsch-Desrousseaux-Pochart, 34, rue de Bassano, F-75008 Paris (FR).	(81) Designated States: CN, JP, KR, US, European patent (AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE). Published <i>With international search report.</i>	
(54) Title: PROCESS OF MANUFACTURING EVOH/POLYESTER BISTRETCHED FILM AND THE FILM THUS OBTAINED (57) Abstract The instant invention relates to a process for manufacturing a multilayer film, such as PET/binder/EVOH/binder/PET, comprising the steps of: (i) coextruding a raw film; (ii) simultaneously stretching it; and (iii) heat treating it at a temperature comprised between 170 °C and 250 °C, and during a heat treatment time comprised between 3 and 8 sec.		

031356 U.S. PTO
10/757837

011504

FOR THE PURPOSES OF INFORMATION ONLY

Codes used to identify States party to the PCT on the front pages of pamphlets publishing international applications under the PCT.

AL	Albania	ES	Spain	LS	Lesotho	SI	Slovenia
AM	Armenia	FI	Finland	LT	Lithuania	SK	Slovakia
AT	Austria	FR	France	LU	Luxembourg	SN	Senegal
AU	Australia	GA	Gabon	LV	Latvia	SZ	Swaziland
AZ	Azerbaijan	GB	United Kingdom	MC	Monaco	TD	Chad
BA	Bosnia and Herzegovina	GE	Georgia	MD	Republic of Moldova	TG	Togo
BB	Barbados	GH	Ghana	MG	Madagascar	TJ	Tajikistan
BE	Belgium	GN	Guinea	MK	The former Yugoslav	TM	Turkmenistan
BF	Burkina Faso	GR	Greece		Republic of Macedonia	TR	Turkey
BG	Bulgaria	HU	Hungary	ML	Mali	TT	Trinidad and Tobago
BJ	Benin	IE	Ireland	MN	Mongolia	UA	Ukraine
BR	Brazil	IL	Israel	MR	Mauritania	UG	Uganda
BY	Belarus	IS	Iceland	MW	Malawi	US	United States of America
CA	Canada	IT	Italy	MX	Mexico	UZ	Uzbekistan
CF	Central African Republic	JP	Japan	NE	Niger	VN	Viet Nam
CG	Congo	KE	Kenya	NL	Netherlands	YU	Yugoslavia
CH	Switzerland	KG	Kyrgyzstan	NO	Norway	ZW	Zimbabwe
CI	Côte d'Ivoire	KP	Democratic People's	NZ	New Zealand		
CM	Cameroon		Republic of Korea	PL	Poland		
CN	China	KR	Republic of Korea	PT	Portugal		
CU	Cuba	KZ	Kazakhstan	RO	Romania		
CZ	Czech Republic	LC	Saint Lucia	RU	Russian Federation		
DE	Germany	LI	Liechtenstein	SD	Sudan		
DK	Denmark	LK	Sri Lanka	SE	Sweden		
EE	Estonia	LR	Liberia	SG	Singapore		

PROCESS OF MANUFACTURING EVOH/POLYESTER
BISTRETCHED FILM AND THE FILM THUS OBTAINED

5

Field of the invention

The invention relates to a process of manufacturing a EVOH/Polyester (coextruded) bistretched film. The invention also relates to such a film, showing high barrier properties and high mechanical properties.

10

Background of the invention

Extrusion of multilayer films is well-known. Stretching of films, being monolayer or multilayer films, is known as well.

15

The above techniques have been applied to many different types of films, in order to obtain the desired results. One multilayer film that would be of high interest is a film containing a layer of EVOH and a layer of polyester. The layer of EVOH would impart barrier properties such as gas barrier properties against oxygen, carbon dioxide, helium, aroma and flavors, etc... The layer of polyester would impart mechanical properties, such as Young's modulus, tensile strength, heat resistance, clarity, etc... Since these two types of polymers are not compatible, a coextrusion binder is necessary. Thus, one would look for films having one layer of EVOH, one layer of coextrusion binder and one layer of polyester.

20

25

The process of manufacturing such films is, however, very delicate. Especially, stretched films are very difficult to manufacture, since the layer of EVOH is very difficult to stretch, especially to bistretch. As a matter of fact, polyester, being nearly amorphous at the exit of the extruder, can be easily stretched, either simultaneously or sequentially, and high quality film is easily obtained. The EVOH layer on the contrary, has a high tendency to crystallise under heat and stress, so that a sequential stretching is not possible: stretching in MD direction forms a crystalline EVOH layer, that breaks whilst the attempt to stretch in TD

30

35

direction. Simultaneous stretching is thus required. Eventually, the stretched films are heat set, so as to develop the mechanical properties of the film, thanks to the polyester layer.

5 Thus, there is a need for a method that would allow the production of valuable multilayer EVOH/PET films, but would also be cost effective and easy to carry out.

JP-A-55139263 discloses a process where a multilayer film is bistretched then heat treated at a temperature below the
10 melting point of the EVOH material. This, however, is not satisfactory, since the melting point of all EVOH grades suitable for high barrier applications is below 190°C, preferably below 180°C, most preferably below 170°C. Such low heat-set temperatures are not appropriate to maintain the
15 thermo-mechanical strength of the outer polyester layers, resulting from the biaxial stretching. This would lead to a high thermal shrinkage and the high mechanical strength of polyester film would be lost at higher temperatures.

JP-A-63272548 discloses a process for manufacturing a
20 laminate stretched film comprising coextruding a 5-layer film, especially polyester/binder/EVOH/binder/polyester, simultaneously bistretching it, and heat-treating it under the following conditions:

$$0.5 \leq X \leq 5$$

25 $\text{EVOH melting point} + 40^\circ\text{C} \leq T \leq \text{polyester melting point}$

$$8 \text{ sec} \leq t \leq 25 \text{ sec}$$

where:

X = total polyester thickness/EVOH thickness

T = heat treatment temperature

30 t = heat treatment time

Examples given in this document provide simultaneous bistretching at a temperature of 90°C, at ratios of 3.3 x 3.3, followed by heat treatment at a temperature of about 230°C, for a duration of about 15 sec. The respective
35 polymers are PET, EVOH and modified PET or modified EVA resin as a binder. The thicknesses (unstretched/stretched) of the EVOH and PET layers are respectively about 54 µm/5 µm and about 130 µm/12 µm, giving a X value of about 2.5. The EVOH

melting point is about 180 (leading to on heatset temperature of 220°C for EVOH mp + 40°C) while PET melting point is about 260°C. At this high heatset temperatures, the film starts to crystallize and gets brittle very fast. Heatset times higher than 8 seconds will definitely lead to film breakage in the oven and a loss of mechanical properties, especially elongation, due to high crystallization and brittleness thereof. Additionally, running a film line with such high hold-up times in the heatset zones would require either very long heatset zones in the oven or very low line speeds. This is economically not suitable for such a product.

Summary of the invention

The object of the present invention is a process where the heat treatment is carried out during a time below 8 sec., and at a temperature preferably below 220°C.

The resulting film shows enhanced properties, especially when the ratio of thicknesses of polyester to EVOH is higher than 5.

Detailed description of the invention

The polyester used in the invention is any polyester where the major part of it is comprised of any aromatic repeating ester units. The term polyester in this invention refers to a polymer that is obtained by condensation polymerization of an aromatic dicarboxylic acid such as terephthalic acid or 2,6-naphthalene dicarboxylic acid and of an aliphatic glycol such as ethylene glycol, 1,4-butanediol or 1,4-cyclohexane dimethanol. These polymers, in addition to being homopolymers, may also be copolymers having a third component or several components. In this case, the dicarboxylic acid component may be, for example, isophthalic acid, phthalic acid, terephthalic acid, 2,6-naphthalene dicarboxylic acid, 4,4'-diphenyldicarboxylic acid, adipic acid, sebacic acid, decanedicarboxylic acid and 1,4-cyclohexane dicarboxylic acid; the oxycarboxylic acid component can be, for example, p-oxybenzoic acid and the glycol component can be, for example, ethylene glycol, diethylene glycol, triethylene glycol, propylene glycol,

butanediol, neopentyl glycol, 1,4-cyclohexane dimethanol, polyethylene glycol and polytetramethylene glycol.

Examples of such polyesters are polyethylenenaphthalate (PEN), polybutyleneterephthalate (PBT), polyethylene-
5 terephthalate (PET), the latter PET being the preferred polyester.

Mixtures are also possible, optionally with another polymer different from a polyester. The intrinsic viscosity of the polyester that is used in the invention may vary from
10 e.g. 0.45 to e.g. 0.7, measured in phenoltetrachlorethane at 30°C. The MW may vary within broad limits, e.g. between 10000 to 30000 g/mol.

The binder material is any material that is adhesive and allow the polyester and EVOH layers to show adhesion, with
15 either adhesive rupture or cohesive rupture. The skilled man will choose the binder thanks to its general knowledge or thanks to routine tests.

Examples of such binders include modified polyolefins, polyacrylates, polyurethanes, polyesters, etc.

20 Examples of binders are the following (co)polymers, grafted with maleic anhydride or glycidyl methacrylate, in which the grafting rate is for example from 0.01 to 5% by weight:

- PE, PP, copolymers of ethylene with propylene, butene,
25 hexene, octene, butadiene, EPR, EPDM, containing, for example, 35 to 80% by weight ethylene, as well as any styrene-based block copolymers such as SBS, SIS, SEBS, and the like;

- ethylene and vinyl acetate (EVA) copolymers containing
30 up to 40% by weight vinyl acetate;

- ethylene and alkyl (meth)acrylate copolymers containing up to 40% by weight alkyl(meth)acrylate;

- ethylene and vinyl acetate (EVA) and alkyl (meth)acrylate copolymers, containing up to 40% by weight
35 comonomers.

Further examples of binders are the following (co)polymers, in which ethylene represents preferably at

least 60% by weight and where the termonomer represents, for example, 0.1 to 10% by weight of the copolymer:

- ethylene/alkyl (meth)acrylate or methacrylic acid/maleic anhydride or glycidyl methacrylate copolymers;

- ethylene/vinyl acetate/maleic anhydride or glycidyl methacrylate copolymers;

- ethylene/vinyl acetate/alkyl (meth)acrylate or methacrylic acid/maleic anhydride or glydicyl methacrylate copolymers.

10 The term "alkyl (meth)acrylate" stands for C1 to C6 alkyl, such as methyl, ethyl, butyl and 2-ethylhexyl methacrylates and acrylates. Moreover, these polyolefins can also be cross-linked using any suitable process or agent (di-epoxy diacid, peroxy, etc.)

15 Still further examples of binders include

- grafted copolymers constituted by at least one mono-amino oligomer of polyamide and of an alpha-mono-olefin (co)polymer grafted with a monomer able to react with the amino functions of said oligomer;

20 Mixtures thereof are also envisaged. The molecular weight of these binders can also vary greatly, as those skilled in the art will understand.

Examples of such binder compositions are provided in the following patents, this list not being exclusive:

25 FR-A-2,291,225, FR-A-2,132,780, EP-A-0,210,307, EP-A-0,033,220, EP-A-0,266,994, EP-A-0,171,777, EP-A-0,342,066, EP-A-0,218,665, USP-4,758,477, USP-4,762,890, USP-4,966,810, USP-4,452,942, USP-3,658,948, USP-5,217,812, all of which being incorporated herein by reference.

30 A preferred binder is an anhydride-modified ethylene vinyl acetate copolymer.

The term "EVOH" as used in the instant invention aims at designing ethylene/vinyl alcohol copolymers containing for example from 15 to 80, preferably 20 to 50 mol% of ethylene.

35 A preferred EVOH contains more than 30 mol%, especially more than 40 mol% of ethylene. This kind of copolymer is well-known in the art, and can be obtained, for example, by saponification of an ethylene/vinyl acetate copolymer, with a

saponification degree of generally more than 90%, most preferably more than 95%. A third monomer can be present, in an amount not adversely hindering the barrier properties. It is permissible to blend other components keeping within a range of not adversely hindering the barrier properties. The blending component should preferably be not more than 40 weight %, more preferably not more than 30 weight %, most particularly not more than 20 weight %.

Such blending resins, for example, include the above ethylene-vinyl alcohol copolymers having an ethylene content different from those which are used as main components, or ethylene-vinyl alcohol copolymers having an ethylene content greater than those which are used as main components, or their partially saponified products, polyamide type polymers, polyester type polymers, ethylene-vinyl ester copolymers, ethylene-aliphatic unsaturated aliphatic acid copolymers, ethylene-aliphatic unsaturated aliphatic acid ester copolymers, ionomer resins, styrene-conjugated diene block copolymers, a partially hydrogenated product of said block copolymer, or else these polymers which have been modified, for example, by grafting a monomer having a carboxylic acid group as a polar functional group, copolymers of ethylene and carbon monoxide, or additionally with the vinyl acetate component or a resin wherein at least some of the acetate groups have been converted into hydroxyl groups, and ethylene, propylene resins, and the like, other than those mentioned above; at least one from these should be used.

Various additives, such as UV-agents, stabilizers, sliding agents, antioxidants, fillers, etc. can be added to each component of the layer, in classical amounts.

The coextrusion referred to in the instant invention is any classical extrusion. The extrusion may be of the type T-die, with feeding block, of the blow film type, etc. Also encompassed in the invention is the extrusion, where individual films are produced, which are then laminated on each other so as to build up a raw multilayer film. In fact, any method that produces a raw film is appropriate; preferably this method is coextrusion.

The extrusion temperature of the EVOH may optimized to reach high barrier properties; said temperature may generally be lower than 220°C, preferably between about 180 and 210°C.

The biaxial stretching or bistretching is carried out simultaneously. It includes stretch ratios of 2 to 5, especially 2.5 to 4, in each direction. The ratio MD stretch ratio/TD stretch ratio is generally comprised between 0.5 to 2, especially 0.7 to 1.3. The suitable stretching temperature is comprised between 75 and 130°C, generally about 90°C. The raw film to be stretched can be preheated, if necessary. Preheat temperature can be as high as the stretching temperature; for example preheat can be performed at a temperature of 80°C (for about 10 sec) while stretching is carried out at a temperature of 90°C. Any simultaneous stretching apparatus can be used; preferably polyester stretching apparatuses are used. Examples of simultaneous bistretching apparatus are disclosed in the following US patents, all incorporated herein by reference: 4,675,582; 4,825,111; 4,853,602; 4,922,142; 5,036,262; 5,051,225; 5,072,493 and 5,416,959.

The heat treatment (or heatsetting) referred to in the instant invention is the classical heat treatment carried out for polyesters; classical apparatuses are used like introducing hot air or using infrared lamps, etc. Either "heatset" or "heat treatment" is used in the instant invention, without any distinction.

The raw film shows a total thickness generally between about 10 and 1000 μm , especially between about 50 and 500 μm . The thickness of the polyester layer (total) is generally between about 20 and 950 μm , preferably between 50 and 300 μm . The thickness of the EVOH layer is generally between about 5 and 100 μm , preferably between 10 and 70 μm . The thickness of the binder layer (total) is generally between about 3 and 30 μm , preferably below 10 μm . The ratio thickness of polyester (total) to thickness of EVOH can vary within broad limits; generally, this ratio is above 1, preferably above 5. The resulting bistretched film shows a thickness generally between about 5 and 200 μm , especially

between about 10 and 100 μm . Each layer, polyester, EVOH, binder is present according to relative thicknesses as given above with respect to the raw film. For example, the EVOH layer may have a thickness between 1 and 10 μm .

5 The instant film formed of the various layers can be of various structures and the polyester used can vary from one layer to another. For example, one layer can be obtained from starting products containing scrap material, allowing recycling. Content of scrap is variable within broad limits
10 known to the skilled man. Also, the polyester can have a different nature from one layer to another, or they can be the same. All polyester layers can be comprised of crystalline polyester, or all layer can be comprised of amorphous copolyester, or one layer can be crystalline and
15 the other(s) can be amorphous. For example, the invention provides 5-layer films where one layer is crystalline (i.e. PET) while the other layer is amorphous (i.e. copolyester). This allows to combine specific mechanical properties of crystalline polyester and adhesive properties of the
20 copolyester. It can also be foreseen that each layer may be formed of two or more sub-layers; e.g. the outer polyester layer can be formed of one sub-layer of PET and one sub-layer of copolyester, the first one being in contact with the binder layer. The instant films can be used as individual
25 layers in further multilayer films.

The following examples further illustrate the present invention, but do not limit the scope thereof.

Examples

EVOH:

30 EVOH polymer was delivered by Kuraray Eval Europe GmbH Duesseldorf.

Grade E105U, 44 mol% ethylene content, density 1.14 g/cm³, melt flow index 5.5 g/10min., melt point 165°C.

BINDER (Bynel):

35 The binder is an anhydride-modified ethylene vinyl acetate copolymer.

Density 0.95 g/cm³, melt index 10.9g/10min, melt point 165°C.

PET:

IV: 0.56 dl/g, melt point 256°C.

The polyester polymer was extruded at a temperature of 280°C, the binder polymer at 240°C. The EVOH barrier polymer was extruded at different temperatures (230°C, 205°C and 175°C). Die temperature was at 290°C.

The cast film samples are summarised in table 1. The thickness of the different layers was confirmed via microtome-cuts under the microscope.

Table 1: Cast film samples

Sample	PET	Bynel	EVOH	Bynel	PET	EVOH Extr. Temp. °C
	µm	µm	µm	µm	µm	
A	50	<10	50	<10	50	230
B	55	<10	40	<10	55	230
C	60	<10	30	<10	60	230
D	65	<10	20	<10	65	230
E	70	<10	10	<10	70	230
F	70	<10	10	<10	70	205
G	70	<10	10	<10	70	175

The cast films were cut into pieces of 11.2cm x 11.2cm, preheated at 80-90°C for 10 sec and simultaneously stretched at 90°C at a stretch ratio of 3.3 x 3.3 in 2.2 sec.

The stretched samples were chucked into a frame for preheating at 170°C, 200°C and 230°C for 3, 8, 15 and 20 seconds. The resulting stretched films are summarised in table 2.

Table 2: Simultaneously stretched films

Roll No.	Preheat Temp. [°C]	Stretch Temp. [°C]	Stretch Ratio	Heatset Temp. [°C]	Heatset Time [sec]
1	80	90	3.3 x 3.3	230	20
2	80	90	3.3 x 3.3	230	20
3	80	90	3.3 x 3.3	230	15
4	80	90	3.3 x 3.3	230	15
5	80	90	3.3 x 3.3	230	8
6	80	90	3.3 x 3.3	230	8
7	80	90	3.3 x 3.3	230	3
8	80	90	3.3 x 3.3	230	3
9	80	90	3.3 x 3.3	200	20
10	80	90	3.3 x 3.3	200	20
11	80	90	3.3 x 3.3	200	15
12	80	90	3.3 x 3.3	200	15
13	80	90	3.3 x 3.3	200	8
14	80	90	3.3 x 3.3	200	8
15	80	90	3.3 x 3.3	200	3
16	80	90	3.3 x 3.3	200	3
17	80	90	3.3 x 3.3	170	20
18	80	90	3.3 x 3.3	170	20
19	80	90	3.3 x 3.3	170	15
20	80	90	3.3 x 3.3	170	15
21	80	90	3.3 x 3.3	170	8
22	80	90	3.3 x 3.3	170	8
23	80	90	3.3 x 3.3	170	3
24	80	90	3.3 x 3.3	170	3

This stretch and heatset pattern was applied for film samples D, E, F and G.

To evaluate the influence of the thickness of the EVOH layer on the oxygen barrier, samples A-G were stretched under the same conditions (see above) and heatset at 200°C for 8 sec.

The oxygen barrier was determined with an Ox-tran twin tester. The following table 2 provides the results.

Table 3: Oxygen barrier of the film samples

Sample	Film Thickness [μm]	EVOH Thickness [μm]	EVOH Extrusion Temp.	Oxygen Barrier $\text{cc/m}^2/\text{d}$
A	15	5	230°C	5.1
B	14	4	230°C	7.5
C	15	3	230°C	10.2
D	14	2	230°C	13.3
E	14	1	230°C	21.6
F	13	1	205°C	6.6
G	13	1	175°C	7.3

As expected, the oxygen permeability is increasing from 5 to 21 $\text{cc/m}^2/\text{d}$ with the decreasing thickness of the EVOH barrier layer from 5 to 1 μm . But also the extrusion temperature of the EVOH polymer has a strong influence on the oxygen barrier. By decreasing the extrusion temperature from 230°C to 205°C the oxygen permeability drops from 21 to 6.6 $\text{cc/m}^2/\text{d}$. According to these results, a 5 μm EVOH layer extruded at 230°C offers the same oxygen barrier than a 1 μm EVOH layer extruded at 205°C.

Samples D-G were examined more intensively regarding the influence of heatset time and temperature. The results of the oxygen barrier are summarised in table 4.

Table 4: Oxygen barrier (expressed in $\text{cc/m}^2/\text{d}$) for samples D-G at different heatset conditions (see table 2)

Roll N°	Sample D	Sample E	Sample F	Sample G
1	15.1	25.7	6.2	9.3
3	17.5	26.3	5.3	9.2
5	14.1	27.4	6.6	7.8
7	16.6	28.0	nd	8.9
9	16.2	25.6	5.9	9.5
11	17.0	26.8	5.7	10.0
13	14.5	27.4	5.7	8.2
15	15.0	24.5	5.5	8.9
17	18.1	25.7	6.3	8.5
19	20.5	28.4	5.9	7.2
21	16.5	21.6	5.7	7.2
23	14.5	23.2	7.3	8.5

20 nd: not determined

Like in table 3, the oxygen permeability is increasing with decreasing EVOH layer thickness and decreasing with lower EVOH extrusion temperature.

However, it can be said that there is no dependence of the oxygen barrier on heatset time and temperature. High oxygen barrier properties may however be obtained with a lower EVOH extrusion temperature, such as about 205°C.

Mechanical properties, i.e. modulus, tensile strength, force at 3% elongation (F3), force at 5% elongation (F5) and elongation at break, along the MD (machine direction) and the TD (transverse direction) were determined with an Instron equipment at room temperature.

For samples A-G, the mechanical properties at heatset conditions of 200°C and during 8 sec, are listed on the following table 5.

For samples D-G, the mechanical properties at all heatset conditions (see table 3), are listed on the following pages on table 6-9.

Table 5: Mechanical properties of samples A-G

Sample	Modulus [N/mm ²]		F3 [N/mm ²]		F5 [N/mm ²]		Tensile [N/mm ²]		Elongation [%]	
	MD	TD	MD	TD	MD	TD	MD	TD	MD	TD
A (15μm)	3610	3563	83	81	97	95	174	166	111	116
B (14μm)	3768	3770	87	86	101	100	182	181	105	117
C (15μm)	3934	3783	90	86	105	102	204	203	110	130
D (14μm)	3862	3731	90	86	105	103	205	190	113	122
E (15μm)	4032	3809	92	89	109	105	201	209	97	131
F (13μm)	3690	3620	84	83	96	95	162	150	112	96
G (13μm)	3741	3726	85	85	98	99	170	154	94	84

Table 6: Mechanical Property Data Sample D (Thickness = 14 μ m)

Roll N°	Modulus		Modulus		F3		F5		Tensile		Tensile		Elonga-	
	MD	TD	MD	TD	MD	TD	MD	TD	MD	TD	MD	TD	tion MD	tion TD
	[N/mm ²]	[N/mm ²]	[N/mm ²]	[N/mm ²]	[N/mm ²]	[N/mm ²]	[N/mm ²]	[N/mm ²]	[N/mm ²]	[N/mm ²]	[N/mm ²]	[N/mm ²]	[%]	[%]
1	3981	4140	87	87	93	93	172	173	127	127	127	127	116	116
3	4005	4064	87	88	95	97	181	184	117	117	117	117	114	114
5	4099	4229	87	87	95	94	183	177	109	109	109	109	114	114
7	4099	4363	86	92	95	101	181	194	106	106	106	106	106	106
9	3896	4025	86	88	97	97	190	188	106	106	106	106	122	122
11	3935	3986	85	85	93	93	179	172	110	110	110	110	109	109
13	4116	4276	88	89	96	97	200	173	121	121	121	121	101	101
15	4120	4208	89	89	98	98	195	192	111	111	111	111	114	114
17	4158	3849	89	85	100	95	200	201	110	110	110	110	126	126
19	4145	4156	89	88	100	99	203	203	114	114	114	114	118	118
21	4051	4074	88	89	98	99	219	208	126	126	126	126	114	114
23	3967	4251	87	93	96	103	212	227	121	121	121	121	118	118

Table 7: Mechanical Property Data Sample E (Thickness = 14 μ m)

Roll N°	Modulus		F3		F5		Tensile		Tensile		Elonga-	
	MD	TD	MD	TD	MD	TD	MD	TD	MD	TD	tion MD	tion TD
	[N/mm ²]	[N/mm ²]	[N/mm ²]	[N/mm ²]	[N/mm ²]	[N/mm ²]	[N/mm ²]	[N/mm ²]	[N/mm ²]	[N/mm ²]	[%]	[%]
1	4541	4770	92	91	99	103	179	193	106	104		
3	4316	4497	92	94	102	103	201	207	118	110		
5	4655	4712	97	96	104	104	200	209	107	112		
7	4488	4508	94	94	102	102	183	198	86	107		
9	4528	4420	95	94	104	103	207	208	113	116		
11	4307	4393	91	92	100	103	198	208	106	121		
13	4421	4344	92	91	101	99	202	192	121	116		
15	4173	4521	89	94	98	104	205	209	129	111		
17	4316	4357	94	94	105	107	212	229	116	116		
19	4186	4269	91	92	101	103	206	196	116	116		
21	4136	4023	87	89	97	100	197	213	102	120		
23	4171	4082	92	90	102	100	207	203	115	119		

Table 8: Mechanical Property Data Sample E (Thickness = 13 μ m)

Roll N°	Modulus MD [N/mm ²]	Modulus TD [N/mm ²]	F3 MD [N/mm ²]	F3 TD [N/mm ²]	F5 MD [N/mm ²]	F5 TD [N/mm ²]	Tensile MD [N/mm ²]	Tensile TD [N/mm ²]	Elonga- tion MD [%]	Elonga- tion TD [%]
1	3416	3688	73	77	78	82	139	155	113	118
3	3742	3388	78	73	82	76	143	133	109	124
5	3603	3334	75	73	78	77	143	123	109	116
7	3211	3516	71	75	75	79	118	130	98	76
9	3247	3580	71	77	77	83	142	145	103	106
11	3577	3476	77	74	83	80	148	135	99	88
13	3199	3608	71	77	76	83	140	147	118	91
15	3216	3590	71	77	76	84	140	141	111	82
17	3799	3729	78	84	78	86	193	190	113	107
19	3745	3619	79	82	85	82	183	183	107	117
21	3655	3609	77	76	83	82	188	178	107	95
23	3436	3294	75	72	82	77	151	144	102	125

Table 9: Mechanical Property Data Sample E (Thickness = 13 μ m)

Roll N°	Modulus MD [N/mm ²]	Modulus TD [N/mm ²]	F3 MD [N/mm ²]	F3 TD [N/mm ²]	F5 MD [N/mm ²]	F5 TD [N/mm ²]	Tensile MD [N/mm ²]	Tensile TD [N/mm ²]	Elonga- tion MD [%]	Elonga- tion TD [%]
1	3818	3971	82	85	89	92	163	173	123	119
3	3996	3868	84	83	91	89	165	159	113	107
5	3891	3853	82	82	88	88	163	157	109	105
7	3710	4008	81	84	87	82	159	162	115	87
9	3614	3926	80	84	87	93	154	168	105	93
11	3577	3845	80	84	88	92	154	167	106	107
13	3653	3748	80	82	88	89	162	155	118	105
15	3674	3803	80	82	88	89	151	161	99	97
17	3898	3831	83	81	91	89	205	186	120	100
19	3560	3530	79	78	87	85	155	157	107	120
21	3677	3677	79	79	86	86	187	194	122	133
23	3425	3536	76	79	82	85	153	153	115	103

18

From the above results, it can be concluded, that a heatset time between 3 and 8 seconds is best suited to obtain a stabilised polyester barrier film with useful mechanical properties and a high oxygen barrier.

5 It can also be concluded that the invention allows processing of EVOH copolymer with high ethylene content while still obtaining very good barrier (oxygen barrier) properties; this is surprising since it is generally admitted that high ethylene content and high barrier property are
10 antinomic.

The invention was described with reference to a preferred embodiment. However, many variations are possible within the scope of the invention.

CLAIMS

- 1.- A process for manufacturing a multilayer film, comprising at least a layer of EVOH, a layer of a coextrusion binder, and a layer of polyester, comprising the steps of:
- (i) coextruding a raw film;
 - (ii) simultaneously bistretching it; and
 - (iii) heat treating it at a temperature comprised between 170°C and 250°C, and during a heat treatment time below 8 sec.
- 2.- The process according to claim 1, where the heat treatment time is comprised between 3 and 8 sec.
- 3.- The process according to claim 1 or 2, where the heat treatment temperature is below 220°C.
- 4.- The process according to any one of claims 1 to 3, where the heat treatment temperature is comprised between 180 and 210°C.
- 5.- The process according to any one of claims 1 to 4, where the ratio (total thickness of polyester of the raw film)/(thickness of EVOH of the raw film) is above 5.
- 6.- The process according to any one of claims 1 to 5, where the EVOH extrusion temperature is below 220°C.
- 7.- The process according to any one of claims 1 to 6, where the simultaneously bistretching is performed at a temperature comprised between 75 and 130°C.
- 8.- The process according to any one of claims 1 to 7, where the simultaneously bistretching is performed at stretching ratios of from 2 to 5, preferably of from 2.5 to 4 in each direction.

20

9.- The process according to any one of claims 1 to 8, where the raw film is preheated before the bistretching step (ii).

5 10.- The process according to any one of claims 1 to 9, where the multilayer film is a 5-layer film polyester/binder/EVOH/binder/polyester.

10 11.- The process according to any one of claims 1 to 10, where the EVOH contains more than 30 mol% of ethylene.

12.- The process according to any one of claims 1 to 11, where the polyester is PET.

15 13.- The process according to any one of claims 1 to 12, where the binder is an anhydride-modified ethylene vinyl acetate copolymer.

INTERNATIONAL SEARCH REPORT

International Application No.

PCT/EP 99/03758

A. CLASSIFICATION OF SUBJECT MATTER

IPC 6 B29C55/02 B32B27/30 B32B27/08

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 6 B29C B32B

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	<p>DATABASE WPI Section Ch, Week 9315 Derwent Publications Ltd., London, GB; Class A17, AN 93-120881 XP002081464 & JP 05 057856 A (SUMITOMO BAKELITE CO), 9 March 1993 (1993-03-09) abstract</p>	1-13
A	<p>DATABASE WPI Section Ch, Week 8920 Derwent Publications Ltd., London, GB; Class A18, AN 89-148542 XP002081465 & JP 63 272548 A (KURARAY CO LTD), 10 November 1988 (1988-11-10) cited in the application abstract</p>	1-13

-/--



Further documents are listed in the continuation of box C.



Patent family members are listed in annex.

* Special categories of cited documents:

"A" document defining the general state of the art which is not considered to be of particular relevance

"E" earlier document but published on or after the international filing date

"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)

"O" document referring to an oral disclosure, use, exhibition or other means

"P" document published prior to the international filing date but later than the priority date claimed

"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention

"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone

"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.

"A" document member of the same patent family

Date of the actual completion of the international search

6 September 1999

Date of mailing of the international search report

15/09/1999

Name and mailing address of the ISA

European Patent Office, P.B. 5818 Patentlaan 2
NL - 2280 HV Rijswijk
Tel. (+31-70) 340-2040, Tx. 31 651 epo nl,
Fax: (+31-70) 340-3016

Authorized officer

Attalla, G

INTERNATIONAL SEARCH REPORT

International Application No

PCT/EP 99/03758

C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	DATABASE WPI Section Ch, Week 7743 Derwent Publications Ltd., London, GB; Class A13, AN 77-76863Y XP002081466 & JP 52 110781 A ((MISD)), 17 September 1977 (1977-09-17) abstract ---	1-13
A	DATABASE WPI Section Ch, Week 8721 Derwent Publications Ltd., London, GB; Class A18, AN 87-146481 XP002081467 & JP 62 083144 A (MITSUBISHI MONSANTO KK), 16 April 1987 (1987-04-16) abstract ---	1-13
A	EP 0 118 227 A (TOYO SEIKAN KAISHA LTD) 12 September 1984 (1984-09-12) claim 1 ---	1-13
A	US 4 900 612 A (SATO KOICHIRO ET AL) 13 February 1990 (1990-02-13) column 7, line 41 - line 51; example 6 -----	1-13

INTERNATIONAL SEARCH REPORT

Information on patent family members

International Application No

PCT/EP 99/03758

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
JP 5057856 A	09-03-1993	NONE	
JP 63272548 A	10-11-1988	JP 2102026 C JP 8013505 B	22-10-1996 14-02-1996
JP 52110781 A	17-09-1977	JP 58041178 B	10-09-1983
JP 62083144 A	16-04-1987	NONE	
EP 0118227 A	12-09-1984	JP 1797974 C JP 4071701 B JP 59143616 A US 4675219 A ZA 8400867 A	12-11-1993 16-11-1992 17-08-1984 23-06-1987 27-03-1985
US 4900612 A	13-02-1990	JP 1925391 C JP 6057444 B JP 62158043 A JP 1771518 C JP 4065781 B JP 61270155 A EP 0230344 A EP 0202954 A	25-04-1995 03-08-1994 14-07-1987 30-06-1993 21-10-1992 29-11-1986 29-07-1987 26-11-1986

THIS PAGE BLANK (USPTO)